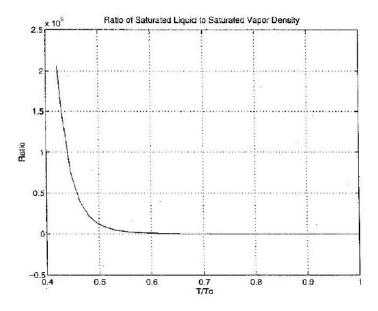
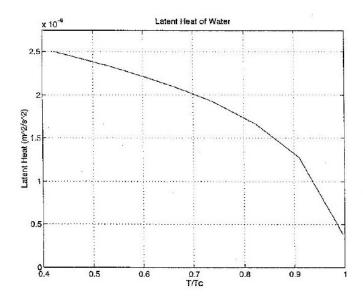
Solution to Problem 100A:

- The temperature and pressure at the triple point of water: $T_t = 273^{\circ}K$; $p_t = 611Pa$
- The temperature and pressure at the critical point of water: $T_c = 647^{\circ}K$; $p_t = 22.1MPa$



- \bullet The temperature and pressure at the triple point of oxygen: $T_t = 54.3^\circ K$; $p_t = 150 Pa$
- \bullet The temperature and pressure at the critical point of oxygen: $T_c=155^{\circ}K$; $p_t=5.08MPa$
- It tends to zero as shown below for water.



• It tends to zero linearly as shown below for water.

